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CASTOR OIL CONVERSION TO BIODIESEL: A PROCESS SIMULATION STUDY

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ABSTRACT

The aim of this study is to highlights the importance to shift from the use of traditional fossil fuels to biodiesel as a clean energy source. A simulation study has been conducted using ASPEN HYSIS software for the biodiesel production form castor oil. The simulation was run and the properties of the produced biodiesel were highlighted. The optimum conditions resulted in 88 % conversion.

Keywords: biodiesel, castor oil, simulation, yield, transesterification.

1. INTRODUCTION

The global economy and socio-economic development are intensively depending on the energy sector. The use of energy in the world is increasing overtime. Every year, the rate of energy is increasing 23%; the main cause of this increase is the overpopulation. The main source of energy is the fossil fuel as the Energy Information Administration (EIA) mentioned with 86.4% of primary energy used in the world gets from fossil fuel (EIA 2019). This main source of energy is one of dangerous sources that led to air pollution and greenhouse gas emissions. Different combustion products are being released to our atmosphere such as volatile organic components (VOCs), carbon monoxide (CO), nitrogen oxide (NO_x) and particulate matter (PM). The Ozone which is the main component of smog is produced within the presence of NO_x and VOC_s combined with any source of heat. The CO₂ produced by the combustion of fossil fuel is equal to 21.3 billion tones/year, after the need of natural process, the rest equals to 10.65 billion tones CO₂ in the atmosphere air (Prasad and Dhanya 2011). The second main source of energy is coal, whose combustion produces Sulphur dioxide (SO₂) which affect the local pollution. The methane emissions are caused by the biomass combustion which emits 40% of gross carbon dioxide and 38% of tropospheric Ozone (Olivier and Peters 2020). There are many natural problems inflicted by these emissions such as the global warming. On the other hand, there are another natural issues such as deforestation and water scarcity due to the overpopulation that are also affected by these emissions.

In the recent years, a holistic thinking to reduce the air pollution by switching from the fossil fuels to other clean energy sources(Al-Akraa, Mohammad *et al.* 2011, Al-Akraa, Mohammad *et al.* 2012, Al-Akraa, Mohammad *et al.* 2015, Al-Akraa, Mohammad *et al.* 2015, Al-Akraa, Mohammad *et al.* 2015, Al-Akraa, Mohammad *et al.* 2015, Al-Akraa, Mohammad *et al.* 2015, Al-Akraa, Mohammad *et al.* 2015, Al-Akraa, Mohammad *et al.* 2017, Al-Akraa, Mohammad *et al.* 2017, Al-Akraa, Asal *et al.* 2018, Al-Akraa, Asal *et al.* 2018, Al-Akraa, Mohammad *et al.* 2018, Mohammad, Al-Akraa *et al.* 2018, Al-Akraa, Asal *et al.* 2019, Al-Akraa, Asal *et al.* 2019, Al-Akraa, Asal *et al.* 2019, Al-

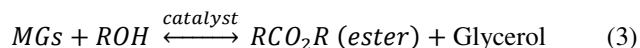
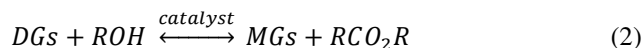
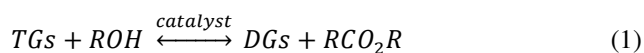
Akraa, Ohsaka *et al.* 2019, Asal, Al-Akraa *et al.* 2019, Asal, Al-Akraa *et al.* 2019, Abdulhalim, Asal *et al.* 2020, Abuzaied, Asal *et al.* 2020, Al-Akraa, Al-Qodami *et al.* 2020, Al-Akraa, Al-Qodami *et al.* 2020, Al-Akraa and Mohammad 2020, Al-Akraa, Salama *et al.* 2021, Al-Qodami, Alalawy *et al.* 2021, Ali, Asal *et al.* 2021, Asal, Mohammad *et al.* 2021). Biofuel is also an alternative for fossil fuels and is considered as a friendly fuel to the nature in limiting the air emissions and improving the air quality. The biofuel is derived from biomass, for example wood residual, forest residual, agriculture products, industrial and urban residues. It involves mainly bioethanol, biodiesel, and biogas and bio hydrogen. (Dhanya 2011).

The formulation of biodiesel yield is obtained from the literature data and hence utilized in Aspen HYSYS process simulation. Aspen HYSYS is a process modeling tool that is used for optimizing, conceptual designing, planning, managing, and monitoring the performance of production, processing, refining, and separating industries. It is a main section of Aspen Tech's aspen ONE® Engineering applications. It has a massive application for chemical engineers in process simulation. It was recognized as a very instinctive and simple process simulator used in gas and oil refining industries. It has very spontaneous proficiencies that include high cooperative flow diagram for navigating and structuring processes through great simulations.

Simulating the technical process is important for designing, analyzing, developing, and optimizing the process. It is mostly applied in chemical plants and power stations. Simulating a process serves as a model-based representation of physical, chemical, and biological unit operations. Simulating a process is used to illustrate the ultimate effects of alternate conditions and followed actions.

2. EXPERIMENTAL PART

ASPEN HYSYS version 10 was used to simulate the production of biodiesel by the transesterification process (Equations 1-3) (Keera, El Sabagh *et al.* 2018):



The input design, conditions, compositions and material streams, respectively, for the methanol and the oil are shown in Figure-1 and Figure-2.

methanol	
Stream Name	methanol
Vapour / Phase Fraction	0.0000
Temperature [C]	75.00
Pressure [kPa]	202.6
Molar Flow [kgmole/h]	3.121
Mass Flow [kg/h]	100.0
Std Ideal Liq Vol Flow [m3/h]	0.1257
Molar Enthalpy [kJ/kgmole]	-2.335e+005
Molar Entropy [kJ/kgmole-C]	88.34
Heat Flow [kJ/h]	-7.288e+005
Liq Vol Flow @Std Cond [m3/h]	0.1256
Fluid Package	<i>Basis-1</i>
Utility Type	

Figure-1. The input conditions, compositions and material streams for methanol.

oil		oil	
Comments	Dynamics	Comments	Dynamics
Stream Name	oil		
Vapour / Phase Fraction	0.0000		
Temperature [C]	60.00		
Pressure [kPa]	101.3		
Molar Flow [kgmole/h]	1.129		
Mass Flow [kg/h]	1000		
Std Ideal Liq Vol Flow [m3/h]	1.092		
Molar Enthalpy [kJ/kgmole]	-1.871e+006		
Molar Entropy [kJ/kgmole-C]	8488		
Heat Flow [kJ/h]	-2.113e+006		
Liq Vol Flow @Std Cond [m3/h]	1.093		
Fluid Package	Basis-1		
Utility Type			

Figure-2. The input conditions, compositions and material streams for the oil.

3. RESULTS AND DISCUSSIONS

The formulation of biodiesel yield is obtaining from literature data and hence utilized in Aspen Hysys process simulation. The process in Hysys is composed of two streams; one of the oil and the other one is the methanol enters to a Continuously Stirred Tank Reactor (CSTR) where the transesterification reaction takes place, then the vapor stream and the liquid one pass through a

total condenser distillation to separate the oil used (Triolein) from the biodiesel (Methyl-Oleate) and the glycerol. The top stream which contains the biodiesel and glycerol enters to the settling tanks which separate the glycerol from the biodiesel due to the difference in their densities. Figure-3 shows the biodiesel production simulation process and Figure-4 shows the output compositions and materials streams for the biodiesel.

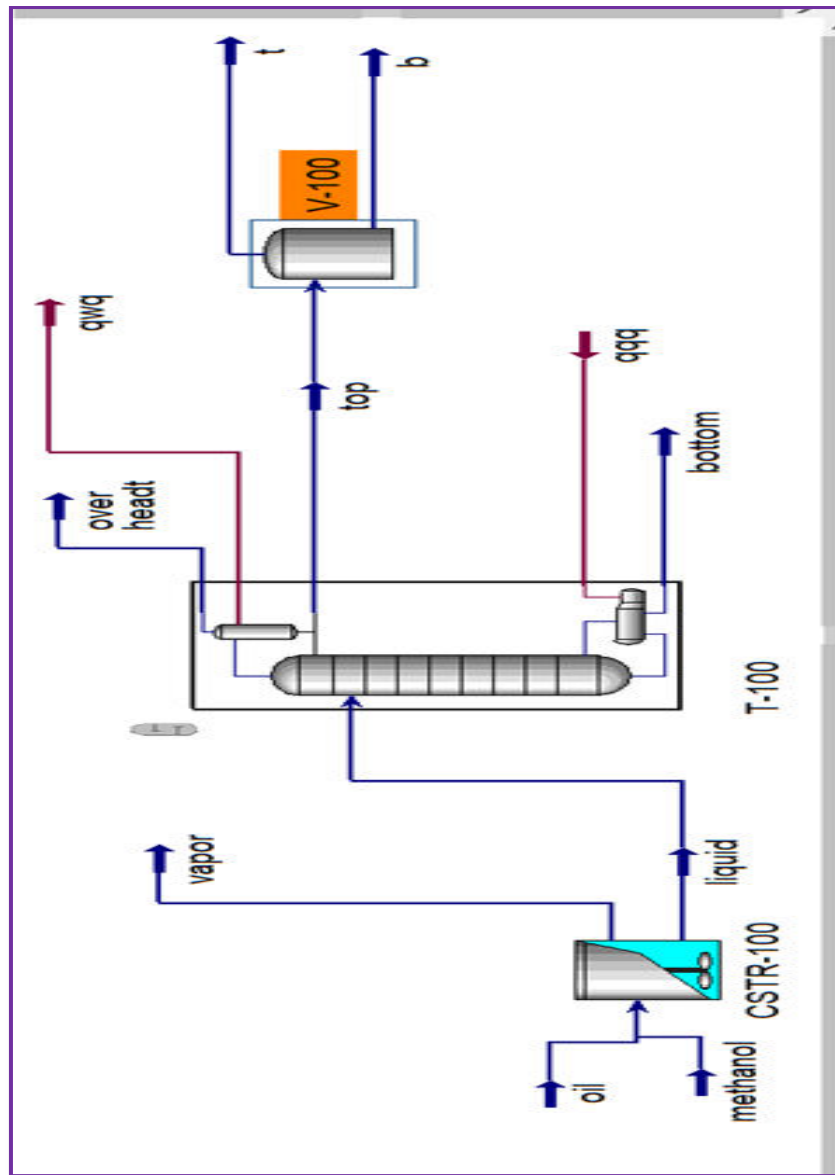


Figure-3. The simulated biodiesel production process.

b		b	
Streams	Dynamics	Streams	Dynamics
Stream Name	b		
Vapour / Phase Fraction	0.0000		
Temperature [C]	304.1		
Pressure [kPa]	101.3		
Molar Flow [kgmole/h]	0.8213		
Mass Flow [kg/h]	223.6		
Std Ideal Liq Vol Flow [m3/h]	0.2520		
Molar Enthalpy [kJ/kgmole]	-5.364e+005		
Molar Entropy [kJ/kgmole-C]	779.5		
Heat Flow [kJ/h]	-4.405e+005		
Liq Vol Flow @Std Cond [m3/h]	0.2414		
Fluid Package	Basis-1		
Utility Type			

Mole Fractions	
NaOH	0.0000
Methanol	0.0024
Triolein	0.0000
Glycerol	0.1156
M-Oleate	0.8820

Figure-4. The output conditions, compositions and material streams for the produced biodiesel.



4. CONCLUSIONS

To decrease pollutant emissions to atmosphere, biodiesel can be used instead of the traditional fossil fuels. This study aimed to simulate the biodiesel production process from castor oil using Aspen Hysys software. The optimum case with the highest conversion (88 %) was demonstrated.

REFERENCES

- Abdulhalim A. S., Y. M. Asal, A. M. Mohammad and I. M. Al-Akara. 2020. Ni-Au anodic nano-electrocatalyst for direct glucose fuel cells. *International Journal of Electrochemical Science*. 15: 3274-3282.
- Abuzaied M. M., Y. M. Asal, A. M. Mohammad and I. M. Al-Akara. 2020. Enhanced glucose electrooxidation at Ni-Cu binary oxide nanocatalyst. *International Journal of Electrochemical Science*. 15(3): 2449-2457.
- Al-Akara I. M. 2017. Efficient electro-oxidation of formic acid at Pd-MnOx binary nanocatalyst: Optimization of deposition strategy. *International Journal of Hydrogen Energy*. 42(7): 4660-4666.
- Al-Akara I. M., B. A. Al-Qodami and A. M. Mohammad. 2020. Effect of the electrodeposition potential of platinum on the catalytic activity of a Pt/GC catalyst toward formic acid electro-oxidation. *International Journal of Electrochemical Science*. 15: 4005-4014.
- Al-Akara I. M., B. A. Al-Qodami, M. S. Santosh, R. Viswanatha, A. K. Thottoli and A. M. Mohammad. 2020. Tuning the activity and stability of platinum nanoparticles toward the catalysis of the formic acid electrooxidation. *International Journal of Electrochemical Science*. 15: 5597-5608.
- Al-Akara I. M., Y. M. Asal and A. M. Arafa. 2018. Fabrication of MnOx/MWCNTs-GC nanocatalyst for oxygen evolution reaction. *International Journal of Electrochemical Science*. 13(9): 8775-8783.
- Al-Akara I. M., Y. M. Asal and S. A. Darwish. 2019. A simple and effective way to overcome carbon monoxide poisoning of platinum surfaces in direct formic acid fuel cells. *International Journal of Electrochemical Science*. 14(1): 8267-8275.
- Al-Akara I. M., Y. M. Asal and A. A. Khalifa. 2019. A promising modification of Pt surfaces with CNTs for decreasing poisoning impact in direct methanol fuel cells. *International Journal of Electrochemical Science*. 14(1): 8276-8283.
- Al-Akara I. M., Y. M. Asal and S. D. Khamis. 2018. Assembling of NiOx/MWCNTs-GC anodic nanocatalyst for water electrolysis applications. *International Journal of Electrochemical Science*. 13(10): 9712-9720.
- Al-Akara, I. M., Y. M. Asal and A. M. Mohammad 2019. Facile synthesis of a tailored-designed Au/Pt nanoanode for enhanced formic acid, methanol, and ethylene glycol electrooxidation. *Journal of Nanomaterials*. 2019.
- Al-Akara I. M. and A. M. Mohammad. 2020. A spin-coated TiOx/Pt nanolayered anodic catalyst for the direct formic acid fuel cells. *Arabian Journal of Chemistry*. 13(3): 4703-4711.
- Al-Akara I. M., A. M. Mohammad, M. S. El-Deab and B. E. El-Anadouli. 2011. Electrooxidation of formic acid at platinum-gold nanoparticle-modified electrodes. *Chemistry Letters*. 40(12): 1374-1375.
- Al-Akara I. M., A. M. Mohammad, M. S. El-Deab and B. E. El-Anadouli. 2012. Development of tailor-designed gold-platinum nanoparticles binary catalysts for efficient formic acid electrooxidation. *International Journal of Electrochemical Science*. 7(5): 3939-3946.
- Al-Akara I. M., A. M. Mohammad, M. S. El-Deab and B. E. El-Anadouli. 2015. Electrocatalysis by design: Synergistic catalytic enhancement of formic acid electro-oxidation at core-shell Pd/Pt nanocatalysts. *International Journal of Hydrogen Energy*. 40(4): 1789-1794.
- Al-Akara I. M., A. M. Mohammad, M. S. El-Deab and B. E. El-Anadouli. 2015. Electrocatalysis by nanoparticle: Enhanced electro-oxidation of formic acid at NiOx-Pd binary nanocatalysts. *Journal of the Electrochemical Society*. 162(10): F1114-F1118.
- Al-Akara I. M., A. M. Mohammad, M. S. El-Deab and B. E. El-Anadouli. 2015. On the catalytic activity of palladium nanoparticles-based anodes towards formic acid electro-oxidation: Effect of electrodeposition potential. *Progress in Clean Energy*. 1: 559-570.
- Al-Akara I. M., A. M. Mohammad, M. S. El-Deab and B. E. El-Anadouli. 2015. On the catalytic activity of palladium nanoparticles-based anodes towards formic acid electro-oxidation: Effect of electrodeposition potential. *Progress in Clean Energy, Volume 1: Analysis and Modeling*: 551-558.
- Al-Akara I. M., A. M. Mohammad, M. S. El-Deab and B. E. El-Anadouli. 2017. Flower-shaped gold nanoparticles: Preparation, characterization, and electrocatalytic application. *Arabian Journal of Chemistry*. 10(6): 877-884.
- Al-Akara I. M., A. M. Mohammad, M. S. El-Deab and B. E. El-Anadouli. 2018. Fabrication of CuOx-Pd nanocatalyst supported on a glassy carbon electrode for enhanced formic acid electro-oxidation. *Journal of Nanotechnology* 2018.
- Al-Akara I. M., A. M. Mohammad, M. S. El-Deab and B. E. El-Anadouli. 2015. Advances in direct formic acid fuel cells: Fabrication of efficient Ir/Pd nanocatalysts for



formic acid electro-oxidation. *International Journal of Electrochemical Science*. 10(4): 3282-3290.

Al-Akraa I. M., T. Ohsaka and A. M. Mohammad. 2019. A promising amendment for water splitters: Boosted oxygen evolution at a platinum, titanium oxide and manganese oxide hybrid catalyst. *Arabian Journal of Chemistry*. 12(7): 897-907.

Al-Akraa I. M., A. E. Salama, Y. M. Asal and A. M. Mohammad. 2021. Boosted performance of NiOx/Pt nanocatalyst for the electro-oxidation of formic acid: A substrate's functionalization with multi-walled carbon nanotubes. *Arabian Journal of Chemistry*. 14(10).

Al-Qodami, B. A., H. H. Alalawy, I. M. Al-Akraa, S. Y. Sayed, N. K. Allam and A. M. Mohammad. 2021. Surface engineering of nanotubular ferric oxyhydroxide goethite on platinum anodes for durable formic acid fuel cells. *International Journal of Hydrogen Energy*.

Ali A. H., Y. M. Asal and I. M. Al-Akraa. 2021. Simulation design studies for biodiesel production from castor oil. *ARPJ Journal of Engineering and Applied Sciences*. 16(16): 1630-1640.

Asal Y. M., I. M. Al-Akraa, A. M. Mohammad and M. S. El-Deab 2019. A competent simultaneously co-electrodeposited Pt-MnOx nanocatalyst for enhanced formic acid electro-oxidation. *Journal of the Taiwan Institute of Chemical Engineers*. 96: 169-175.

Asal Y. M., I. M. Al-Akraa, A. M. Mohammad and M. S. El-Deab. 2019. Design of efficient bimetallic Pt-Au nanoparticle-based anodes for direct formic acid fuel cells. *International Journal of Hydrogen Energy*. 44(7): 3615-3624.

Asal, Y. M., A. M. Mohammad, S. S. A. El Rehim and I. M. Al-Akraa. 2021. Preparation of Co-electrodeposited Pd-Au Nanocatalyst for Methanol Electro-oxidation. *International Journal of Electrochemical Science*. 16: 1-11.

Dhanya. 2011. intechopen. Air Quality and Biofuels.

EIA. 2019. Today in Energy.

Keera S. T., S. M. El Sabagh and A. R. Taman 2018. Castor oil biodiesel production and optimization. *Egyptian Journal of Petroleum*. 27(4): 979-984.

Mohammad A. M., I. M. Al-Akraa and M. S. El-Deab 2018. Superior electrocatalysis of formic acid electro-oxidation on a platinum, gold and manganese oxide nanoparticle-based ternary catalyst. *International Journal of Hydrogen Energy*. 43(1): 139-149.

Mohammad A. M., G. A. El-Nagar, I. M. Al-Akraa, M. S. El-Deab and B. E. El-Anadouli. 2015. Towards improving the catalytic activity and stability of platinum-based

anodes in direct formic acid fuel cells. *International Journal of Hydrogen Energy*. 40(24): 7808-7816.

Olivier J. G. J. and J. A. H. W. Peters. 2020. Trends in global CO₂ and total greenhouse gas emissions: 2019 Report Netherlands Environmental Assessment Agency. 4068.

Prasad S. and M. S. Dhanya. 2011. Air Quality and Biofuels, Environmental Impact of Biofuels. IntechOpen.